

Optical spectroscopy in materials science 6. Practical KK transformation and anisotropy

Kamarás Katalin

MTA Wigner FK

kamaras.katalin@wigner.mta.hu



Budapesti Műszaki és Gazdaságtudományi Egyetem

Optical anisotropy

- χ , ε , n are tensors
- if the medium is not isotropic, there is different response to excitations of different polarization

Polarization	Quantity	Notation	
linear	n'	(linear) birefringence <i>(lineáris) kettőstörés</i>	
linear	n''	(linear) dichroism <i>(lineáris) dikroizmus</i>	
circular	n'	optical rotation <i>optikai forgatás</i>	optical activity <i>optikai aktivitás</i>
circular	n''	Or: circular birefringence <i>cirkuláris kettőstörés</i> circular dichroism <i>cirkuláris dikroizmus</i>	



Optical activity

Optical rotatory dispersion (ORD): different index of refraction for left and right circularly polarized light



polarization plane of linearly polarized light will rotate

Circular dichroism: extinction coefficient different



linearly polarized light changes into elliptical



Optical rotation

angle of rotation:

$$\theta = (n_L' - n_R') \frac{\omega l}{2c} = \alpha d$$

$$\frac{\theta}{l} = \alpha = (n_L' - n_R') \frac{\omega}{2c} = (n_L' - n_R') \frac{\pi}{\lambda}$$

specific rotation (for solutions) :
fajlagos forgatóképesség

$$[\alpha] = \frac{\theta}{lc'}$$

Important!

$\lambda = 589 \text{ nm}$ (Na D-line)

$T = 25 \text{ }^{\circ}\text{C}$

solvent: water

$[c']$: g/100 cm³

$[l]$: dm

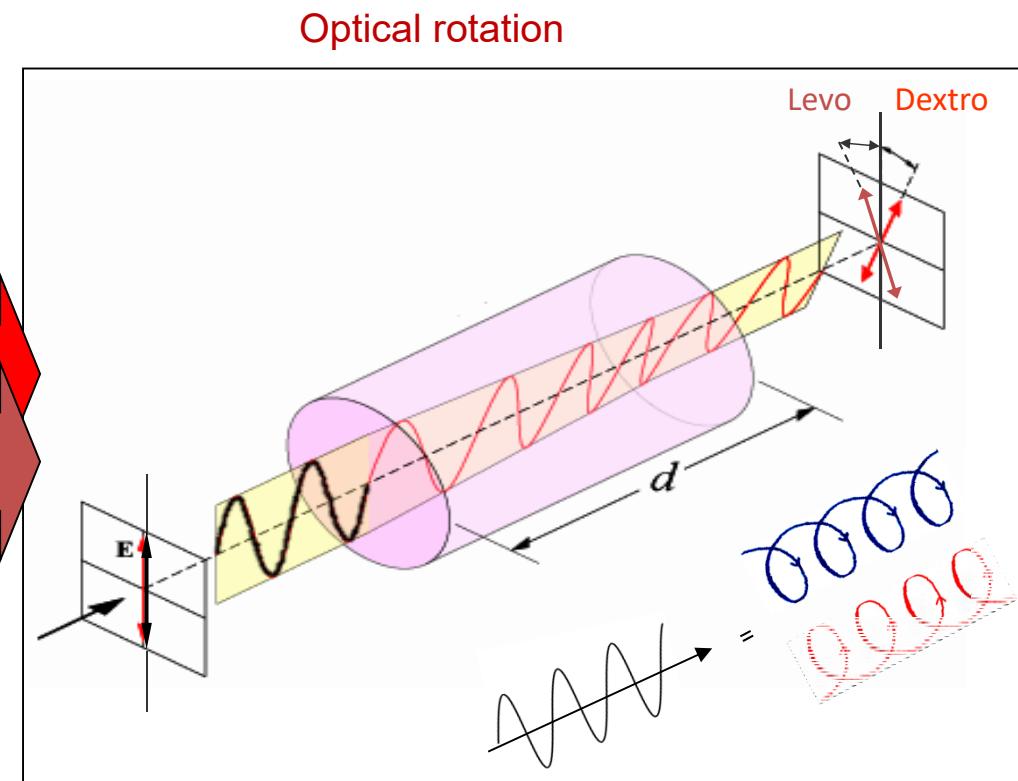
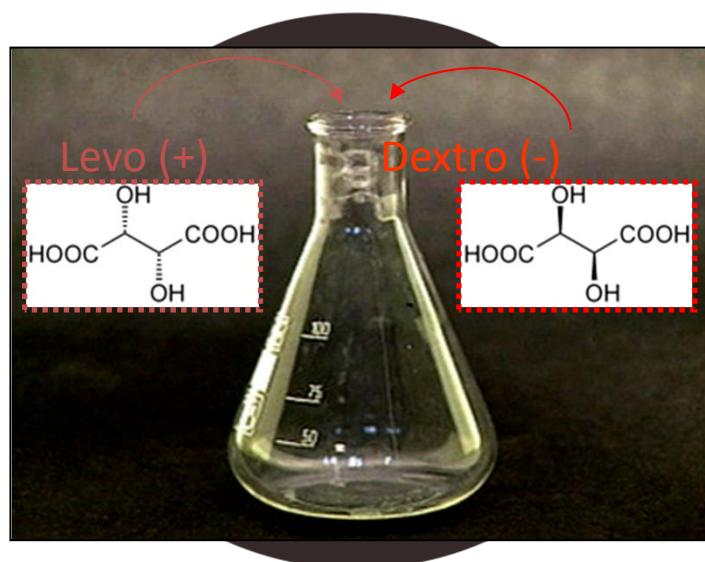
Application: sugar concentration
(health system, food industry)
polarimetry, sacharimetry



Natural optical rotation

Kézsmárki István, BME Fizika Tsz
tartaric acid

Pasteur (1849)



Circular dichroism

Ellipticity: n'' different \longrightarrow absorption different $\longrightarrow \Delta\epsilon \sim \Psi$

$$\frac{\phi}{l} = \Psi = \frac{\pi}{\lambda} (n_L'' - n_R'') \quad \phi \text{ ellipticity, } \Psi \text{ ellipticity created on 1 cm pathlength}$$

specific ellipticity
fajlagos ellipticitás:

$$[\Psi] = \frac{100\phi}{lc'}$$

also used: difference in extinction (absorption) coefficient:
(specific, molar)

$$\Delta\epsilon = \epsilon_L - \epsilon_R \cong n_L'' - n_R''$$
$$\Delta\epsilon(\omega) = \frac{4}{c_0 \log 10} \Psi(\omega)$$

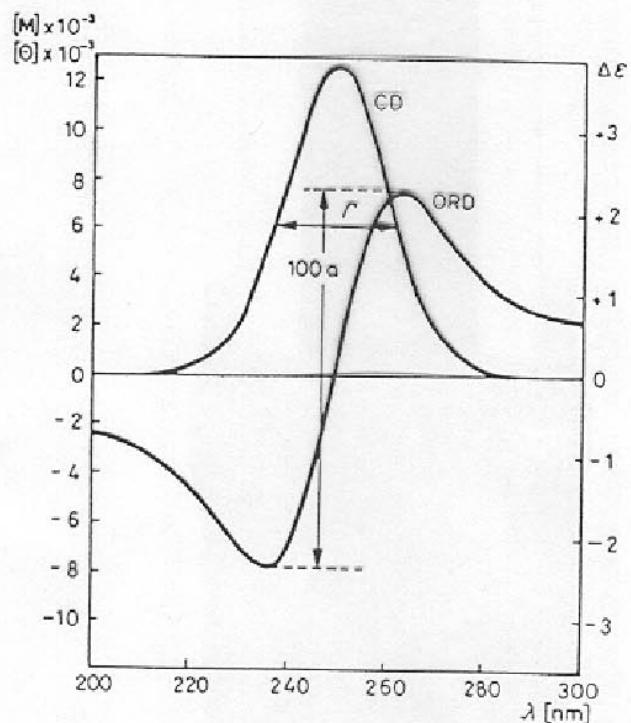
Summarized in one quantity:

$$\Phi = \alpha + i\Psi = \frac{\pi}{\lambda} (n_L - n_R)$$

Definition of R, L and sign is arbitrary!



Cotton effect



Pozitív Cotton-effektusnak megfelelő (idealizált) CD- és ORD-görbepá� ($R = 0,1 \text{ D} \cdot \mu_B$,
 $\Gamma = 25 \text{ nm}, \lambda_{\max} = 250 \text{ nm}$)

$$\Phi = \alpha + i\Psi = \frac{\pi}{\lambda} (\tilde{n}_L - \tilde{n}_R)$$

α and Ψ are related
by Kramers-Kronig relations

Positive Cotton effect: $\Psi > 0$
Negative: $\Psi < 0$

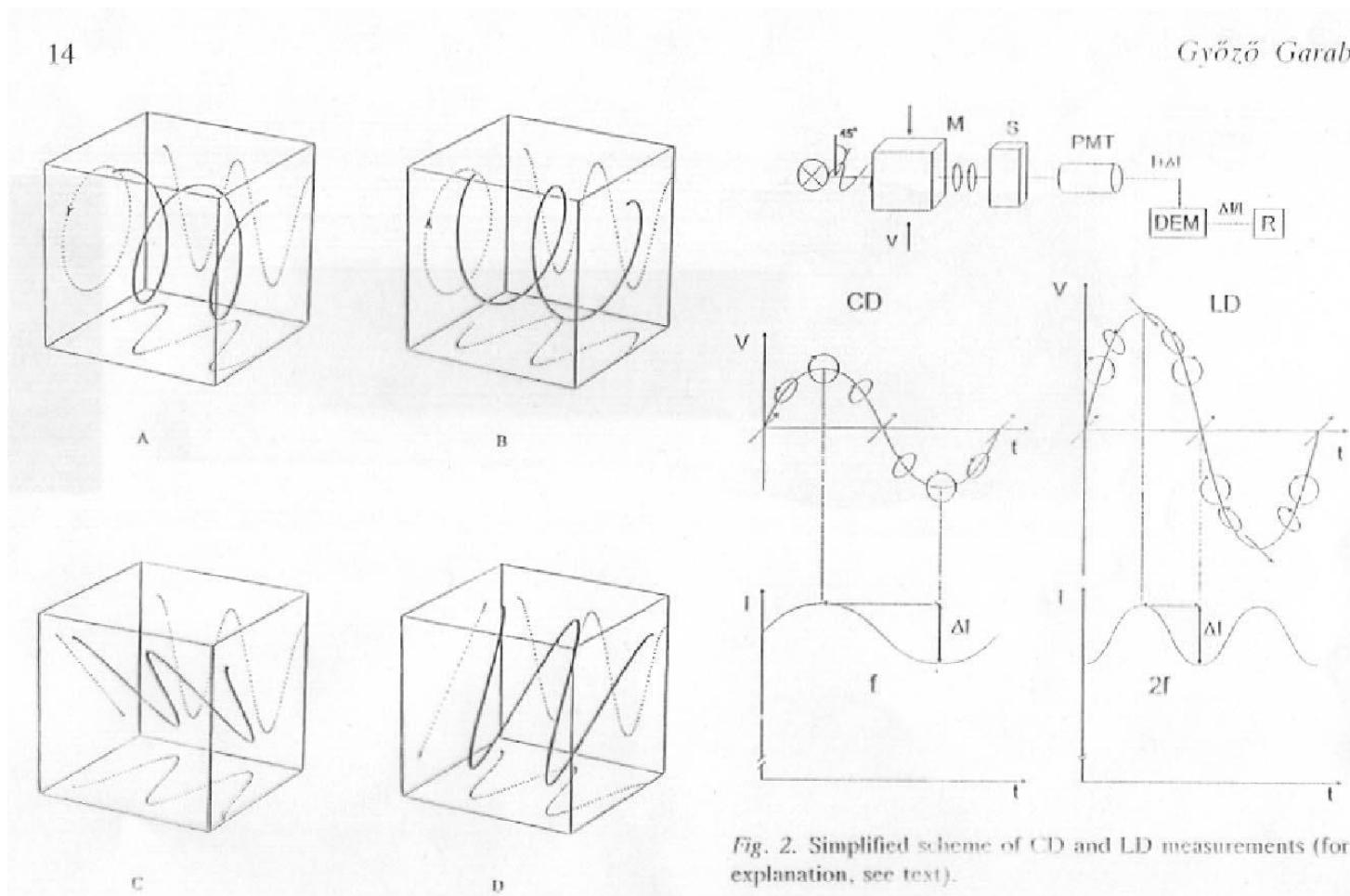


Applications

- information about optical isomers (identical structure, except for rotation)
- resolution can be increased in some cases(sign!)
- magnetic dipole transitions can increase intensity
- ORD: in the whole frequency range
- CD: only around excitations



Measurement: dichrograph – piezoelectric modulator



Garab Győző, SZBK



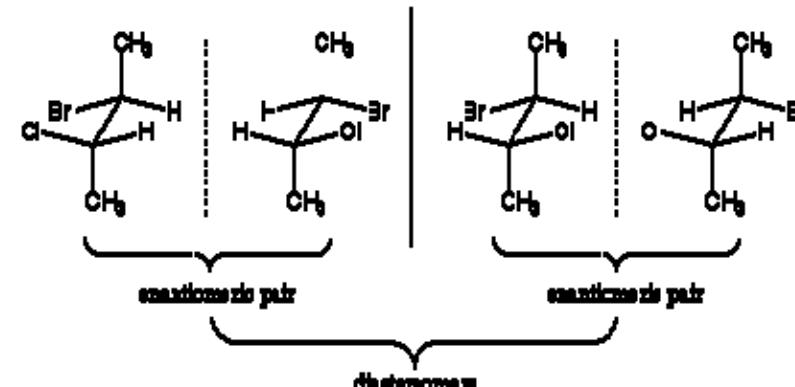
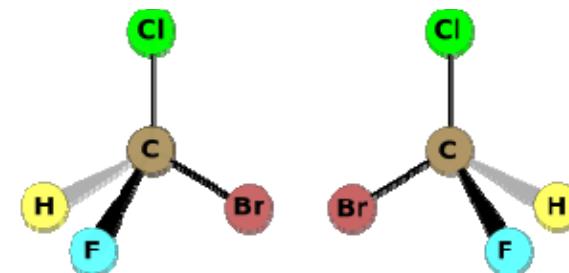
Optical isomerism

Structural condition: chirality
(lack of symmetry element S_n)

Optical isomerism:

- **enantiomers:** can be transformed into each other by reflection across a plane
- **racemic mixture (racém keverék):** mixture of enantiomers in a 1:1 ratio
- **diastereomers:** isomers formed by the reaction of two chiral molecules (*not enantiomers*)

[http://en.wikipedia.org/wiki/Chirality_\(chemistry\)](http://en.wikipedia.org/wiki/Chirality_(chemistry))
enantiomers

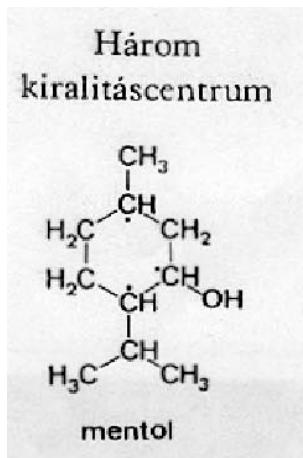


<http://www.sparknotes.com/chemistry/organic3/enantiomersanddiastereomers/section2.rhtml>



Optical isomerism

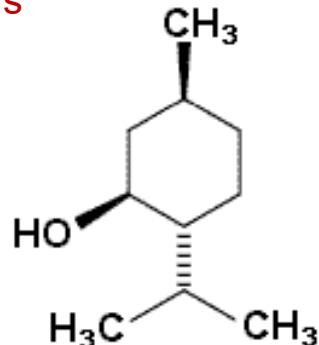
Three chirality centers



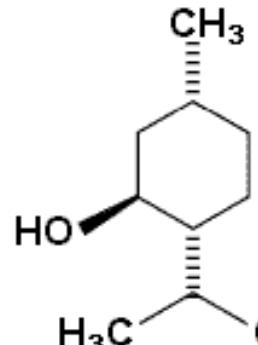
menthol

Kajtár
Márton:
Változatok
négy elemre

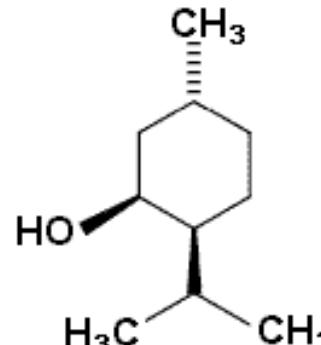
(+)-Menthol



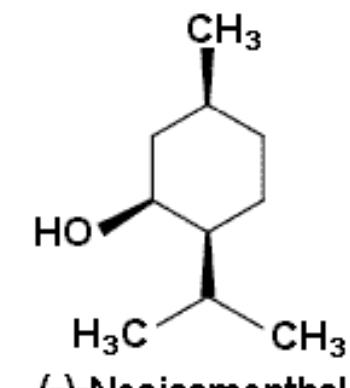
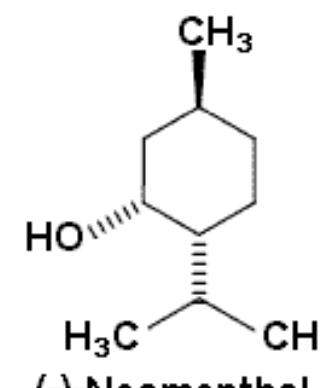
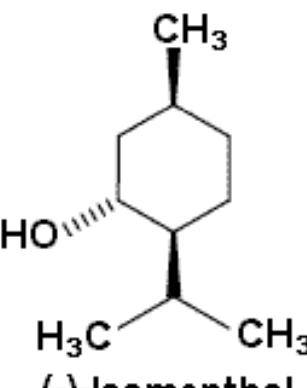
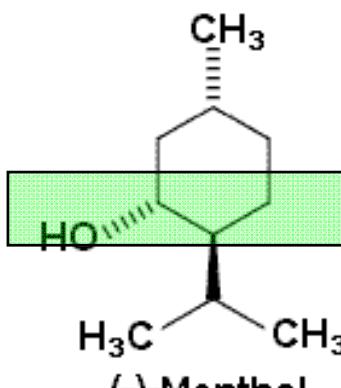
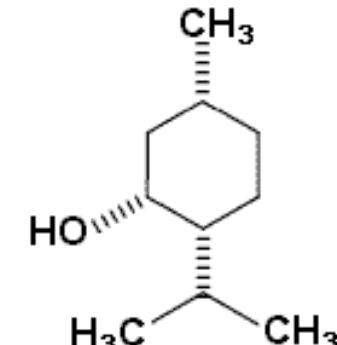
(+)-Isomenthol



(+)-Neomenthol



(+)-Neoisomenthol



<http://en.wikipedia.org/wiki/Image:Menthols.PNG>

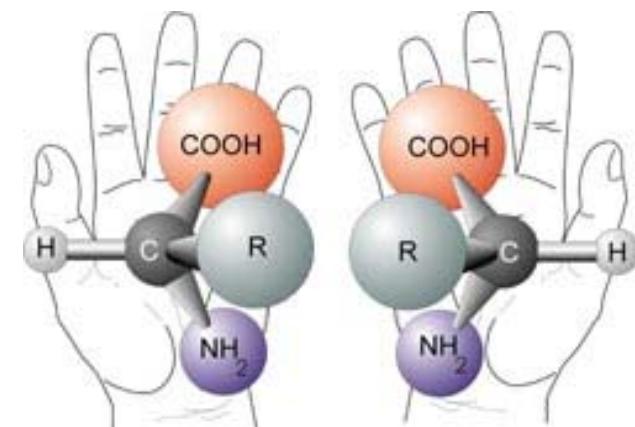


Separation of optical isomers

Separation of optical isomers:
symmetric synthesis → diastereomers

Natural substances (enzymes, amino acids):
only one enantiomer exists
(*nature is chiral*)

Application: age determination
(amino acid racemization)



<http://www.nai.arc.nasa.gov/>



Reflectance spectroscopy

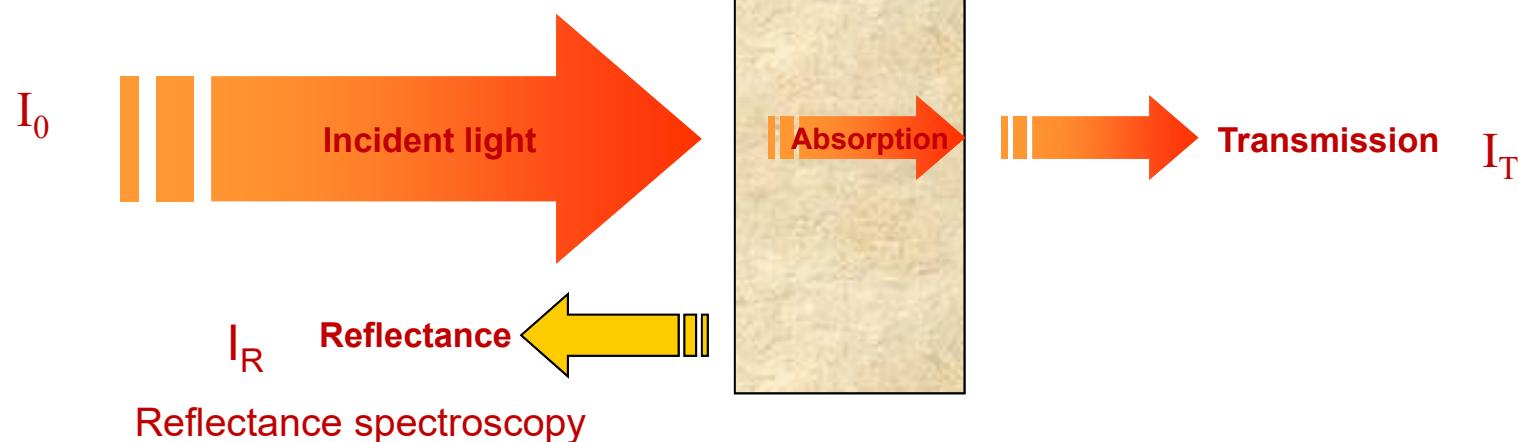
$$I_0 = RI_0 + (1-R)I_0 e^{-\alpha d} + I_A$$

$$I_T \cong 0$$

Fresnel equations for normal incidence:

$$r = \frac{E_r}{E_i} = \frac{n' + in'' - 1}{n' + in'' + 1} = \sqrt{R} e^{i\theta}$$

$$R = \frac{I_R}{I_0} = \frac{(n'-1)^2 + n''^2}{(n'+1)^2 + n''^2}$$



Dispersion relations - reflectance

$$r = \frac{E_r}{E_i} = \frac{n' + in'' - 1}{n' + in'' + 1} = \sqrt{R} e^{i\theta} \quad \ln r = \frac{\ln R}{2} + i\theta$$

11.51 egyenlet hibás!
(-1-szerese)

In r is also a good transfer function, dispersion relations hold
measured quantity: $R(\omega)$

Kramers-Kronig (KK) transformation:

$$\theta(\omega) = -\frac{\omega}{\pi} \text{P} \int_0^\infty \frac{\ln R(\xi)}{\xi^2 - \omega^2} d\xi = \frac{-1}{2\pi} \text{P} \int_0^\infty \ln \left| \frac{\xi + \omega}{\xi - \omega} \right| \frac{d \ln R(\xi)}{d\xi} d\xi$$

Contribution to θ : $\xi - \omega$ small or $d \ln R / d\xi$ large

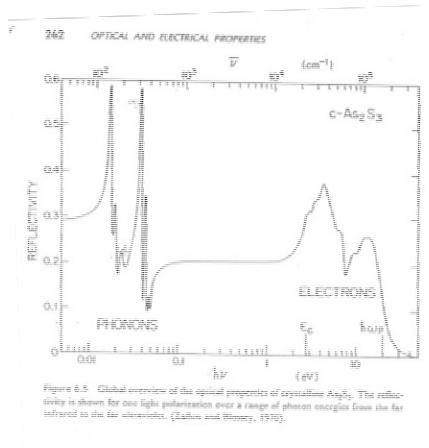


Figure 6.5 Global overview of the optical properties of crystalline As_2S_3 . The reflectivity is shown for one light polarization over a range of photon energies from the far infrared to the far ultraviolet. (Zallou and Werner, 1999).



KK analysis: extrapolations

$\omega \rightarrow 0$:

- known $\sigma(\text{dc})$: fitting, smoothing
- semiconductors, bound electrons: constant R
 $\epsilon''=0$, ϵ' can be calculated from R

$\omega \rightarrow \infty$:

asymptotic limit (free electrons):

$$R \sim \left(\frac{\omega}{\omega_p} \right)^{-4} \quad \text{reflectance of plasma } (\omega > 10^6 \text{ cm}^{-1}, \text{ all electrons excited})$$

intermediate region: $R \sim \left(\frac{\omega}{\omega_p} \right)^{-S}$

S (and even ω_p) arbitrary, because we only want to approximate the integral, $R \sim \omega^{-S}$

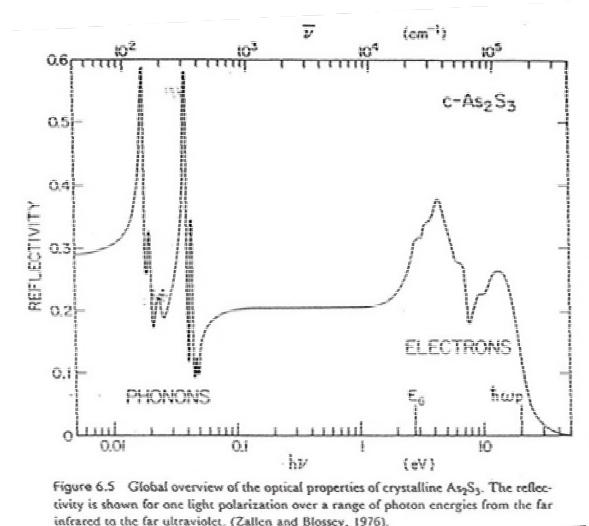


Figure 6.5 Global overview of the optical properties of crystalline As_2S_3 . The reflectivity is shown for one light polarization over a range of photon energies from the far infrared to the far ultraviolet. (Zallen and Blossey, 1976).

KK analysis: extrapolations

$\omega \rightarrow 0$:

- known $\sigma(\text{dc})$: fitting, smoothing
- semiconductors, bound electrons: constant R
 $\epsilon''=0$, ϵ' can be calculated from R

$\omega \rightarrow \infty$:

asymptotic limit (free electrons):

$$R \sim \left(\frac{\omega}{\omega_p} \right)^{-4} \quad \text{reflectance of plasma } (\omega > 10^6 \text{ cm}^{-1}, \text{ all electrons excited})$$

intermediate region: $R \sim \left(\frac{\omega}{\omega_p} \right)^{-S}$

S (and even ω_p) arbitrary, because we only want to approximate the integral, $R \sim \omega^{-S}$

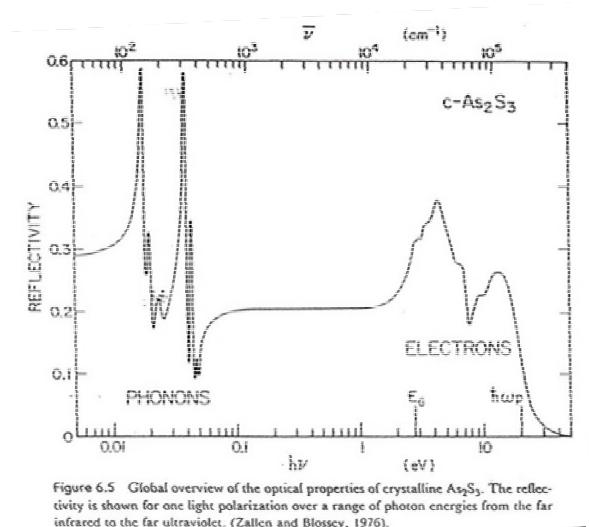
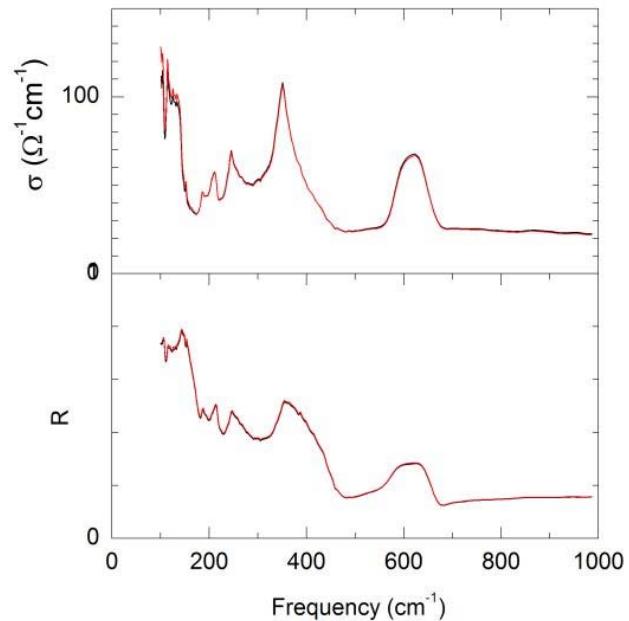


Figure 6.5 Global overview of the optical properties of crystalline As_2S_3 . The reflectivity is shown for one light polarization over a range of photon energies from the far infrared to the far ultraviolet. (Zallen and Blossey, 1976).

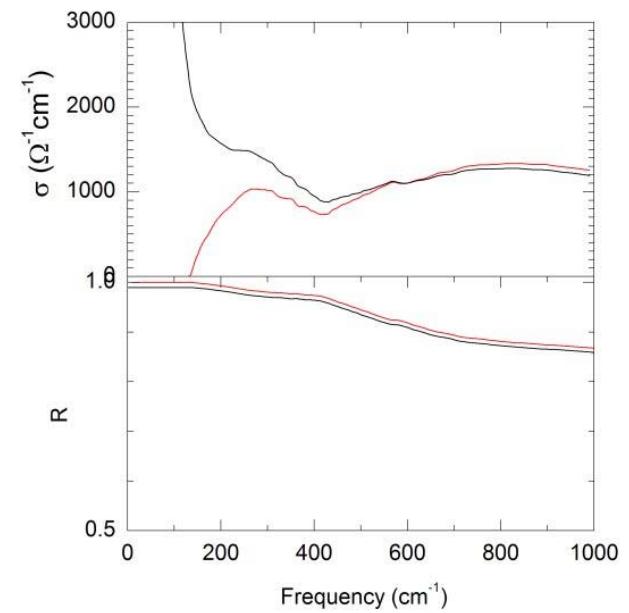
KK analysis: errors

$$R \cong 0, R \cong 1$$

$$n = \frac{1+r}{1-r}$$



Semiconductor



Superconductor



KK analysis: procedure

1. ϵ'' or σ (any extrapolation)
2. starting parameters (Drude-Lorentz ω_p , ω_0 , γ)
3. fit to **measured** quantity (reflectance)
4. KK transformation with extrapolation from model frequency range: only **measured** range

